



# Measurement of Uranium Concentrations in the Soil Samples of Nineveh Province, Iraq Using CR-39 Detector

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**Abstract:** The physiological influence of radiation, which may induce cancer, makes the radioactive field vital to human health. Radon, which is hazardous to both humans and the environment, is deposited in soil via uranium decay. Extended exposure to elevated levels of alpha radiation, primarily radon, causes lung cancer. Thus, research is necessary to observe how the levels of human exposure vary. In evaluating natural exposure radiation, it is essential to control the quantity of radionuclides in the soil by determining the radioactivity level from these sources. Twenty samples of soil from different places in the Nineveh province in northern Iraq were analyzed for uranium content by the CR-39 nuclear track detector. These samples had radon concentrations ranging from 27.221 to 59.407 Bq.m<sup>-3</sup>, and the mean is 38.917 Bq.m<sup>-3</sup>. This mean value is below the level of reference limits 100 Bq.m<sup>-3</sup> of the World Health Organization (WHO). Uranium levels ranged values from 0.129 to 0.281 ppm, with a mean of 0.182 ppm. Findings demonstrate that the natural radiations in the region are within the permissible range, as shown when comparing the results to global averages, the uranium concentration levels fell below the average of global value 2.8 ppm, and below the acceptable limit of 11.7 ppm.

**Keywords:** Nineveh Province, Soil Samples, Uranium Concentrations, CR-39 Detector, Radon Concentrations.

## 1. INTRODUCTION

Soil is a major natural source of radioactivity because of its mineral richness. It also makes people more vulnerable to radiation and helps release radioactive materials into the environment. For these reasons, soil's inherent radioactivity is thought to be a key sign of radioactive contamination [1]. The concentration of radionuclides in soil is the primary indicator of natural background radiation [2]. The natural radioactivity can differ significantly across different types of soil. Natural occurring radioactive elements (NORMs) are present in soil and the environment. These elements are the released nuclear radiation from naturally occurring radionuclide materials. Uranium is one of these radionuclides that is produced by nuclear technology [3]. Similar to other minerals, uranium is an element naturally occurring that has existed since the beginning of the earth. Since each uranium isotope is radioactive, its quantity needs to be controlled [4]. It turns into a

non-radioactive lead as it decays, releasing alpha particles. A progeny, also known as a decay product, is a new radionuclide that occurs along the decay chain. It adds approximately seven times as much radioactivity to the soil as uranium does. In rocks and soil, The closest source of radium and radon is uranium. One of the primary causes of background radiation is radon, a uranium derivative [5]. Because of its chemical and radioactive characteristics, uranium that accumulates in people may have two effects. Humans may suffer negative consequences from consuming large amounts of uranium and the byproducts of its decay. The Harmful effects on humans are caused by high uranium intake and by-products of its decay. Exposure to naturally soluble uranium of 0.1 mg/kg causes chemical temporary kidney damage [6]. Primordial radionuclides enter the soil through the weathering of the earth's crust. It took the radionuclides in the soil by plants up through their roots and passed them on to humans when crops are consumed, resulting in internal

exposure to ionizing radiation. Radionuclides consumed through food are major radiation pathways for health long-term considerations, significantly contributing to average doses of radiation to different human body parts and tissues. The primary uranium and radium deposition sites are the kidneys, liver, and bones [7]. The uranium amount in the soil varies greatly depending on the geological region; it can range from 0.1 to 20 ppm [8], with a global average of 2.8 ppm and 11.7 ppm being the permitted limit [9, 10]. The plastic CR-39 detector was used to measure The amount of uranium present in soil, due to the great sensitivity of CR-39 to alpha particles of low energy. These tracks are examined under a microscope following the etching procedure to magnify them [11]. The goal of the research is to identify the precise uranium content in twenty samples of soil from the Nineveh province in Iraq.

## 2. MATERIALS AND METHODS

Twenty samples of soil were gathered in the Iraqi province of Nineveh from various places as in (Figure 1). The samples were taken 30 cm below the surface. The samples were sieved using a 0.2 mm mesh and baked for 24 hours at 100 °C. After that, samples were placed inside a uranium dosimeter, which was covered with plastic tape to keep radionuclides from escaping into the atmosphere (Figure 2). Utilizing a radioactive uranium dosimeter, solid-state track detector techniques were employed in the measurements. Each container has a 7 cm diameter and a height of 10 cm. It holds one CR-39 detector (1 x 1) cm<sup>2</sup>. Upon arriving at the

detector, alpha particles leave traces. The average concentration of radon is directly correlated with the number of tracks. A sample of 50 g in a cylindrical plastic container (uranium dosimeter) was placed with its back to the CR-39 detector. The sample is 3 cm in height and 7 cm from the detectors to the sample surfaces, closed for 60 days. Detectors were removed after 60 days, and the tracks were revealed by the NaOH etching at normality 6.25 N in the water bath at 70 °C heated. Detectors were cleaned and then tracks were tallied using a microscope at 400x magnification. The following formula is used to calculate track densities [12].

$$\rho = (\text{The tracks number})/(\text{The field of view area}) \quad (1)$$

The track density  $\rho$ (in track.cm<sup>-2</sup>) in sample air space is related to the exposure duration  $T$ (in a day), and then concentrations of radon  $C_{Rn}$ (Bq.m<sup>-3</sup>) can calculated using the relation [13].

$$C_{Rn} = \rho/KT \quad (2)$$

To calculate the radon concentration, the track density of the detector  $\rho$ (track.m<sup>-2</sup>) was used. The radon concentration  $C_{Rn}$ (Bq.m<sup>-3</sup>) during the period of irradiation  $t$ (s) is connected to the density of tracks, or total tracks number per area of the detector's field of view via microscope, utilizing the sensitivity of detector  $K$ (m), as in the equation (2). The sensitivity  $K$  in a unit of (m) is defined as the density of track per exposure unit of (Bq.s.m<sup>-3</sup>). From equation (2), the unit  $K = (\text{track. m}^{-2})/(\text{Bq. s. m}^{-3})$ .

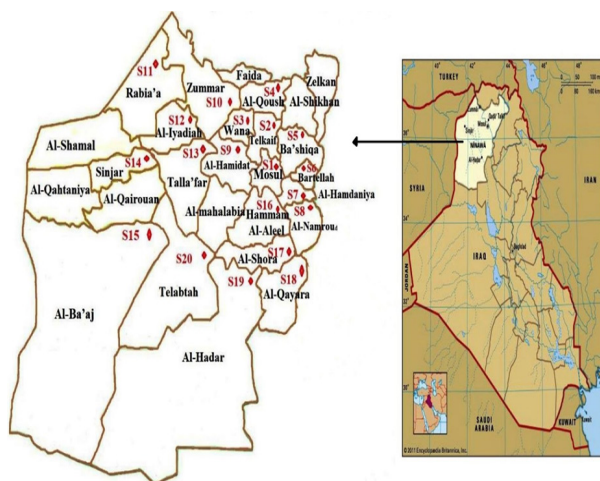


Fig 1. The sitemap of the samples.

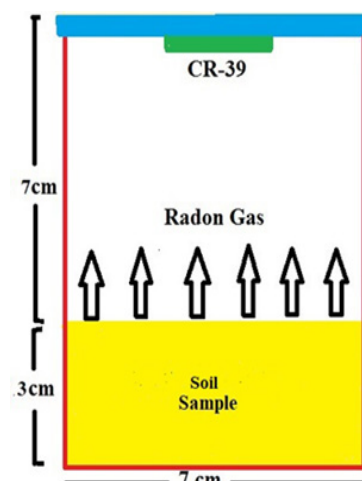


Fig 2. The sealed-can technique (uranium dosimeter).

Since the unit of Bq is equal to decay/s and each decay leaves a track on the detector then 1Bq = 1track/s, then K will be in unit of (m). The Sensitivity K is given traditionally, in track/cm<sup>2</sup> per Bq.d/m<sup>3</sup>. To convert K from the definition in equation (3) to traditionally, the units (cm, d) must converted to (m, s), where 1m = 100 cm, 1day = 86400s, then substituting the values of m, s in equation (3) produce

$$(m) = (\text{track.cm}^{-2}10^{-4})/(\text{Bq.d} \frac{1}{86400} \cdot \text{m}^{-3}).$$

Then,  $(m) = 8.64(\text{track.cm}^{-2})/(\text{Bq.d} \cdot \text{m}^{-3})$ .

Divided by 8.64 to convert to m or m must be multiplied by 8.64 to convert to track.cm<sup>-2</sup> per Bq.d.m<sup>-3</sup> or by 0.0864 if K in cm) [14]. K determined using the relation.

$$K = \left(\frac{1}{4}\right)r(2\cos\theta_c - r/R_i) \quad (3)$$

where, the sealed container radius r is 3.5 cm,  $\theta_c$  is the critical angle of the CR-39 detector equal to 35°, and  $R_i$  is the range of alpha particle in air equal to 4.09 cm (for alpha particle emitted from radon with energy 5.49 MeV) [15-17].  $R_i$  is calculated by the relation.

$$R_i = 0.318 E_i^{3/2} \quad (4)$$

K is then equal to 0.684 cm and has the value (0.0591 Traks.cm<sup>-2</sup>.day<sup>-1</sup>/Bq.m<sup>-3</sup>) after multiplying by 0.0864 [18]. The formula determines the sample's radon concentration Rn [19].

$$C_s = \lambda_{Rn} C_{Rn} HT/L \quad (5)$$

where  $C_s$  (Bq.m<sup>-3</sup>) represents the inside sample's radon concentration, the  $C_{Rn}$  (Bq.m<sup>-3</sup>) represents radon in the air of the sample,  $\lambda_{Rn}$  (0.1814 day<sup>-1</sup>) of the sample air space The can's air space height, T (sixty days) is the exposure time, and the radon decay constant, H (7 cm). L (3cm) is the thickness of the sample. The activity of the sample's radon ( $A_{Rn}$ ) was calculated using the formula.

$$A_{Rn} = C_s V \quad (6)$$

where  $A_{Rn}$  represents the sample radon activity. The scan radius is r, and the sample's volume is ( $V = \pi r^2 L$ ) = 115.4X10<sup>-6</sup>m<sup>3</sup> [20]. From the radon activity and the number of radon atoms  $N_{Rn}$ , the uranium

concentration is obtained using the following formula [21].

$$A_{Rn} = \lambda_{Rn} N_{Rn} \quad (7)$$

The uranium atoms number in the sample ( $N_U$ ) is calculated using the equation of secular equilibrium, which states that uranium activity is equal to radon activity.

$$\lambda_U N_U = \lambda_{Rn} N_{Rn} \quad (8)$$

Where  $\lambda_U$  (4.883X10<sup>-18</sup> sec<sup>-1</sup>) is the decay constant of uranium and the equation determines the weight of uranium inside the sample [22].

$$W_U = N_U At_U / N_{avo}. \quad (9)$$

The Avogadro number is  $N_{avo}$ . (6.02X10<sup>23</sup> atom/mol), and the mass number of uranium is represented by  $At_U$ . The concentration of uranium is calculated from relation [23].

$$C_U = W_U / W_s \quad (10)$$

where,  $W_s$  (50 grams) is the sample mass and  $C_U$  (ppm) is the uranium concentration [24].

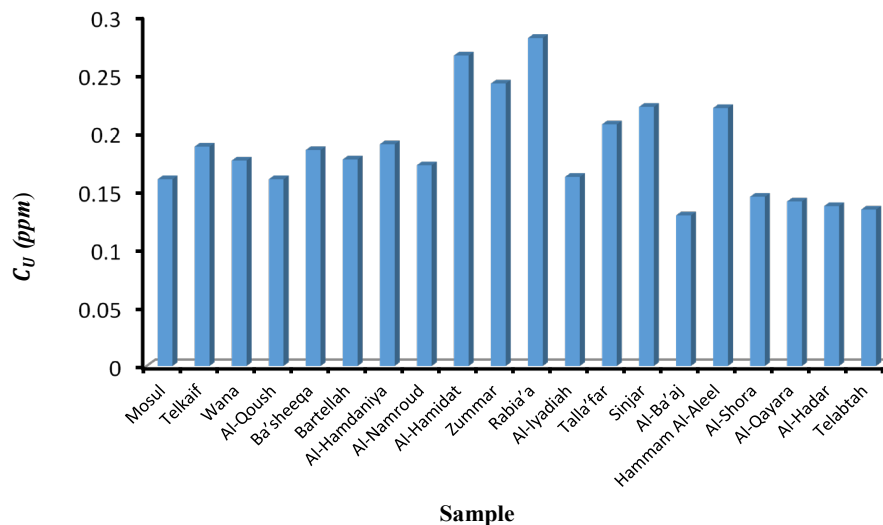
### 3. RESULTS AND DISCUSSION

The results obtained from the present study are presented in Table 1. It was discovered that the range of uranium concentrations was (0.129 - 0.281) ppm, and the mean value was 0.182 ppm as displayed in Figure 3. The estimated radon concentration values in soil samples fall within the range of (27.221 - 59.407) Bq.m<sup>-3</sup>. The uranium content and soil location determine the uranium concentration values in the samples, which vary from one sample to the next. The local geology of the area, along with mineralization, geophysical, and climatological factors, determines the amount of radon gas in the soil and the rate at which it is exhaled. Mountainous areas and near hills have higher concentration values than areas near the river because river flow may reduce concentrations.

Sample S15 Al-Ba'aj had the lowest reported Uranium concentration (0.129 ppm), which is an open soil area without stones far from the river water. While sample S11 Rabia'a had the highest value (0.281 ppm), this location is a large agricultural area

**Table 1.** Soil sample findings related to track density, the radon concentration in the sample air, radon concentration in the soil, and uranium concentration.

| S. No. | Location        | $\rho$ Track / $cm^2$ | $C_{Rn}$ Bq/ $m^3$ | $C_S$ Bq/ $m^3$ | $C_U$ ppm |
|--------|-----------------|-----------------------|--------------------|-----------------|-----------|
| S1     | Mosul           | 120                   | 33.898             | 860.8814        | 0.160     |
| S2     | Telkaif         | 140                   | 39.719             | 1008.709        | 0.188     |
| S3     | Wana            | 131                   | 37.151             | 943.4912        | 0.176     |
| S4     | Al-Qoush        | 119                   | 33.727             | 856.5335        | 0.160     |
| S5     | Ba'shiqa        | 138                   | 39.034             | 991.3179        | 0.185     |
| S6     | Bartellah       | 132                   | 37.322             | 947.8391        | 0.177     |
| S7     | Al-Hamdaniya    | 141                   | 40.061             | 1017.405        | 0.190     |
| S8     | Al-Namroud      | 128                   | 36.295             | 921.7518        | 0.172     |
| S9     | Al-Hamidat      | 198                   | 56.154             | 1426.106        | 0.266     |
| S10    | Zummar          | 180                   | 51.018             | 1295.67         | 0.242     |
| S11    | Rabia'a         | 210                   | 59.407             | 1508.716        | 0.281     |
| S12    | Al-Iyadih       | 121                   | 34.240             | 869.5771        | 0.162     |
| S13    | Talla'far       | 155                   | 43.828             | 1113.059        | 0.207     |
| S14    | Sinjar          | 166                   | 46.909             | 1191.321        | 0.222     |
| S15    | Al-Ba'aj        | 96                    | 27.221             | 691.3138        | 0.129     |
| S16    | Hammam Al-Aleel | 165                   | 46.704             | 1186.103        | 0.221     |
| S17    | Al-Shora        | 108                   | 30.576             | 776.5324        | 0.145     |
| S18    | Al-Qayara       | 105                   | 29.720             | 754.7929        | 0.141     |
| S19    | Al-Hadar        | 102                   | 28.899             | 733.9231        | 0.137     |
| S20    | Telabtah        | 100                   | 28.419             | 721.749         | 0.134     |
|        | Min             | 96                    | 27.221             | 691.3138        | 0.129     |
|        | Max             | 210                   | 59.407             | 1508.716        | 0.281     |
|        | Mean            | 137.5                 | 38.917             | 989.84          | 0.182     |

**Fig 3.** The uranium concentration (ppm) in samples.

**Table 2.** The compression of results with other measurements in Iraq.

| No. | Research     | Location                    | Uranium concentration (ppm) |        |       |
|-----|--------------|-----------------------------|-----------------------------|--------|-------|
|     |              |                             | Min                         | Max    | Mean  |
| 1.  | Present work | Nineveh province -Iraq      | 0.129                       | 0.281  | 0.182 |
| 2.  | [25] 2017    | AL-Hamdaniya-Mosul-Iraq     | 0.313                       | 0.784  | 0.488 |
| 3.  | [26] 2015    | Sulaimani - Iraq            | 1.253                       | 18.225 | 6.029 |
| 4.  | [27] 2014    | Al-Najaf - Iraq             | 0.0935                      | 0.1843 |       |
| 5.  | [28] 2013    | Jalawla'a city-Diyala-Iraq  | 0.719                       | 1.280  |       |
| 6.  | [29] 2022    | Southern Basrah governorate | 0.65                        | 2.67   | 1.382 |
| 7.  | [30] 2023    | Um Qasr district, Iraq      | 0.6                         | 2.2    | 1.3   |

in which organic and chemical fertilizers are widely used. These fertilizers may contain a percentage of radioactive materials, although the percentage is small, but it may contribute to increasing uranium concentrations in this area. The average uranium concentration was 0.182 ppm which is lower than the global average of 11.7 ppm. The compression with recent measurements in Iraq is displayed in Table 2.

Table 2 indicates that the obtained values fall within a certain range of values published for different regions of Iraq. It is far lower than the Sulaimani region, Um Qasr district, and the southern Basrah governorate. Higher than the Al-Najaf area, it lies in the middle of the other measurements in the AL-Hamdaniya region of Mosul and regions of Jalawla'a city-Diyala. The origin of soil types, geochemical makeup, and reported concentrations on dry soil samples all point to the possibility that radioactively rich sandstone is the source of the elevated recorded radionuclide values in the soil sample from that particular area. Therefore, it was concluded that no harmful effects from the radiation and that the population of the research zone is safe from radioactive threats and health concerns coming from the concentrations of uranium in the soil.

#### 4. CONCLUSIONS

The majority of the investigation's uranium-specific activity result values showed that soil samples had less uranium than the average amount found worldwide. Sample S15 Al-Ba'aj had the lowest reported Uranium concentration, which is an open soil area without stones far from the river water. While sample S11 Rabia'a had the highest

value, this location is a large agricultural area in which organic and chemical fertilizers are widely used. These fertilizers may contain a percentage of radioactive materials, although the percentage is small, but it may contribute to increasing uranium concentrations in this area. It was discovered that the uranium content was below both the UNSCEAR-recommended global average of 2.8 ppm and the maximum permissible level of 11.7 ppm. Therefore, it was concluded that no harmful effects from radiation and that the population of the research zone is safe from radioactive threats and health concerns coming from the concentrations of uranium from the soil.

#### 5. CONFLICT OF INTEREST

The authors declare no conflict of interest.

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